Pyrano Heterocycles I. The Syntheses of Isochromans and the Novel Thieno[3,2-c]pyran, Benzothieno[3,2-c]pyran, Benzothieno[2,3-c]pyran and Pyrano[4,3-b]benzofuran Systems

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In a continuing investigation of the synthesis of heterocyclic systems we had occasion to prepare a number of 1substituted isochromans. Isochroman itself has been synthesized in high yield (1,2) by the reaction of phenethyl

alcohol with paraformaldehyde and anhydrous hydrogen chloride, and it has been converted to 1-aminomethylisochroman via the 1-bromo and 1-cyano derivatives (3,4). The direct condensation of benzaldehyde and acetaldehyde with phenethyl alcohol has also been reported (1), to afford 1-phenyl and 1-methylisochromans, in 50% and 7% yields, respectively. It appeared that suitable activation of the aromatic ring would facilitate the synthesis of 1-substituted isochromans.

In practice, we found that 3,4-dimethoxyphenethyl alcohol, 1 (5) reacts readily with aminoacetaldehyde diethylacetal in dioxane with anhydrous hydrogen chloride to give an 82% yield of the hydrochloride of 1-aminomethyl-6,7-dimethoxyisochroman, 2. (See Scheme 1). It was further found that a wide variety of carbonyl derivatives react with 3,4-dimethoxyphenethyl alcohol either in dioxane with hydrogen chloride as catalyst, or, in benzene with p-toluenesulfonic acid as catalyst. Thus, aldehydes, acetals and ketones afforded the 1-substituted and 1,1-disubstituted isochromans 2-11 shown in Table I.

In addition, investigation of the condensation of heterocyclic ethanols with various earbonyl components has afforded a number of novel pyrano-heterocycles (See Scheme II). Thus, reaction of 2-thiopheneethanol (6), benzo[b]-thiophene-2-ethanol (7), benzo[b]-thiophene-3-ethanol (8) and benzofuran-2-ethanol 12 (see Experimental) with miscellaneous carbonyl components has afforded, respectively, 6,7-dihydro-4H-thieno[3,2-c]-pyrans (13-15), 3,4-dihydro-1H-[1]-benzothieno[3,2-c]-pyrans (16,17), the 1,2-dihydro-1H-[1]-benzothieno[2,3-c]-pyran, 18, and the 3,4-dihydro-

1H-pyrano[4,3-b]benzofuran, 19. Data on compounds 13-19 are collected in Table II.

The reactions described, leading to pyranoheterocycles, may be regarded as a variant of the Friedel-Crafts reaction where intramolecular alkylation of an aromatic system by a hemiacetal, hemiketal, or, mixed acetal intermediate affords a pyrano ring.

EXPERIMENTAL

Melting points were taken on a Thomas-Hoover apparatus and are corrected. Analyses were done by Mr. W. Turnbull and staff of Ayerst Research Laboratories. The nmr spectra were recorded on a Varian A-60A instrument. All new compounds gave ir and nmr spectra consistent with their respective structures.

Synthesis of 1-Substituted and 1,1-Disubstituted 6,7-Dimethoxy-isochromans (See Table 1).

Method 1. 1-Aminomethyl-6,7-dimethoxyisochroman (2).

A solution of 3,4-dimethoxyphenethyl alcohol 3 (6) (108 g., 0.59 mole) and aminoacetaldehyde diethylacetal (78 g., 0.59 mole) in dry dioxane (150 ml.) was stirred and cooled to 0° for 1 hour while being saturated with hydrogen chloride. The mixture was kept at 22° for 48 hours and the crystalline precipitate was collected, washed with dioxane, then with ether, and dried to afford the product (126 g., 82%).

Method II. 6,7-Dimethoxy-1-(2'-thienyl)isochroman (9).

A mixture of 3,4-dimethoxyphenethyl alcohol 3 (20 g., 0.11 mole), thiophene-2-carboxaldehyde (15.1 g., 0.13 mole), anhydrous benzene (350 ml.) and p-toluenesulphonic acid (0.55 g.) was heated at reflux in an apparatus fitted with a Dean-Stark trap. The cooled mixture was washed with brine, dried and evaporated. The

(Table I)

1-Substituted and 1,1-Disubstituted-6,7-dimethoxyisochromans

				Dogwoods (h)					ی	~	Ì	
No.	Carbonyl Component	Method (a)	% Yield	Solvent	M.p. °C R ¹	\mathbb{R}^1	\mathbb{R}^2	Formula	Calcd.	Calcd. Found	Caled. Found	Found
7	CH(OEt) ₂ CH ₂ NH ₂	_	82	V	250-255	Н		$C_{12}H_{18}CINO(c)$	55.60	55.85	6.84	7.23
က	CH(OEt)2CH2Br	_	20	B,C	78-80	Н		$C_{12}H_{15}BrO_3(d)$	50.18	50.10	5.26	4.98
4	$MeCO(CH_2)_2COOH$	Н	52	C,D	105-108	Me		$C_{15}H_{20}O_{5}$	64.27	64.40	7.19	7.02
വ	MeCOC ₆ H ₅	Ι	09	C	93-95	Me		$C_{18}H_{20}O_{3}$	76.03	76.19	5.09	7.03
9	$C_6H_5CO(CH_2)_3CI$	I	29	A	125-126	C_6H_5		$C_{20}H_{23}ClO_3$ (e)	69.20	68.98	99.9	6.46
7	C ₅ H ₄ N(3-CHO) (f)	_	75	Ŧ	143-145	Н		$C_{18}H_{19}NO_{7}(g)$	59.83	60.02	5.30	5.28
œ	$C_4H_3O(2\text{-}CHO)$ (f)	П	06	B,F	94-96	Н		$C_{15}H_{16}O_{4}$	69.21	69.34	6.20	6.17
6	C ₄ H ₃ S(2-CH0) (f)	II	80	C,D	78-81	H		$C_{15}H_{16}O_{3}S(h)$	65.21	65.15	5.84	5.64
6	3(0H)C6H4CH0	_	45	D	135-137	Н		$C_{17}H_{18}O_4$	71.31	71.27	6.34	6.23
7	4(H00C)C ₆ H ₄ CH0	-	35	D	140-143	H	4(H00C)C ₆ H ₄	$C_{18}H_{18}O_{5}$	68.78	69.02	5.77	5.79

(a) See Experimental. (b) A, methanol; B, ether; C, hexane; D, benzene; E, 2-propanol; F, petroleum ether (b. p. 30-60°). (c) Anal Calcd. Cl, 13.68; N, 5.40. Found: Cl, 10.18. (f) C₅H₄N = pyridyl; C₄H₃O = furyl; C₄H₃S = thienyl. (g) Anal. Calcd. N, 3.86. (h) Anal. Calcd. S, 11.58. Found: S, 11.58.

(Table II)

Pyrano-Fused Heterocyclic Systems (a)

No.	No. Carbonyl Component % Yield	% Yield	Recrystn. Solvent	$M.p.$ °C R_1 (b)	R ₁ (b)	R ₂ (b)	Formula	Calcd.	C% Found	H Caled.	C% H% S% Calcd. Found Calcd. Found	S% Calcd.	, Found
					6,7	6,7-DIHYDRO-4H-THIENO[3,2-c]PYRANS	3,2~]PYRANS						
5 4	CH(OEt) ₂ CH ₂ NH ₂ MeCOCH ₂ COOEt	60 63	$^{2 ext{-PrOH}}_{ ext{C}_{6} ext{H}_{6}}$	204-206 89-92	Ж	$CH_2NH_2 \cdot HCI$ $CH_2COOH (d)$	$C_8H_{12}CINOS(c) \\ C_{10}H_{12}O_3S$	46.70 56.60	46.46 56.76	5.88	6.15 5.74	15.60 15.03	15.59 15.10
5	N-Methylpiperidone	37	2.PrOH	220-222		MeN $(CH_2)_2$ $(CH_2)_2$	$(CH_2)_2$. $(CH_2)_2$.	51.80	51.57	7.20	96.9	11.50	11.57
				สว	4-DIH	3,4-DIHYDRO-1 <i>H</i> -[1]BENZOTHIENO[3,2-c]PYRANS	ENO[3,2-c]PYRANS						
16	CH(OEt) ₂ (CH ₂) ₂ Cl MeCOCH ₂ COOEt	70 50	МеОН ЕtOH	92-94 149-152	H	(CH ₂) ₂ Cl CH ₂ COOH(d)	$C_{13}H_{13}ClOS(g)$ $C_{14}H_{14}O_{3}S$	61.80 64.11	61.80 63.97	5.18 5.38	5.10 5.42	12.65 12.20	12.64 12.25
					1,2-DIH	1,2-DIHYDRO-1 <i>H</i> -[1]BENZOTHIENO[2,3-<]PYRAN	IENO[2,3-€]PYRAN						
18	$CH(OEt)_2CH_2NH_2$	74	2-PrOH	169.171	H	CH2NH2·maleate	$C_{16}H_{17}NO_{5}S(h)$	57.31	57.37	5.11	5.19	9.56	9.52
					3,4-DII	3,4-DIHYDRO-1 <i>H</i> -PYRANO[4,3-6]BENZOFURAN	-b]BENZOFURAN						
19	19 MeCOCH ₂ COOEt	20	2-PrOH	157-158	Me	$CH_2COOH(d)$	C14H14O4	68.28	68.17	5.73	5.80		

(a) All condensations in this Table were done by Method I, see Experimental. (b) See Scheme II for structural formulae. (c) Anal. Calcd. Cl. 17.30; N, 6.82. Found: Cl. 17.14; N, 6.64. (d) The primary condensation product, the corresponding ethyl ester, was not isolated but was hydrolysed directly to the acid. (e) Obtained as a monohydrate. (f) Anal. Calcd. Cl. 12.80; N, 5.00. Found: Cl. 12.97; N, 5.38. (g) Anal. Calcd. Cl. 14.05. Found: Cl. 13.93. (h) Anal. Calcd. N, 4.18. Found: N, 4.15.

residue was dissolved in benzene and filtered through a column of neutral alumina (Woelm, activity III), and the eluates were concentrated. Addition of hexane afforded the crystalline product, (24 g., 80%).

Benzofuran-2-ethanol (12).

Butyl lithium in hexane (50 ml. of a 2.2 M solution) was added to a solution of benzofuran (11.8 g., 0.1 mole) in ether (200 ml.) during 15 minutes. The mixture was stirred at 22° for 0.5 hour then cooled to 0° while ethylene oxide (4.6 g., 0.105 mole) in ether (50 ml.) was added. After 3 hours at 0° the reaction mixture was washed with brine and the ether evaporated. The residue was purified by chromatography on silica gel. Elution with benzene-chloroform mixtures afforded the product (5.2 g., 45%) as an oil, homogenous by tle, nmr (deuteriochloroform): δ 1.83 (s, 1, 0H), 3.08 (t, J = 6.5 cps, 2, CH_2CH_2OH), δ 3.92 (t, J = 6.5 cps, 2, CH_2CH_2OH), δ 7.3, m, 4, aromatic protons.

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